

# AMEOX® Technology

## for PFAS Destruction...

...in Spent Granular Activated Carbon,  
Brines, & Concentrates.

*"AMEOX processed GAC  
returned to PFAS water  
treatment service  
outperformed the original  
virgin GAC in adsorption  
capacity."*

**RAMBOLL**



# YB Technologies, LLC

Anacortes, WA

- Founded in the Pacific NW (2019)
- A Technology Applications Company providing:
  - Bench/Pilot-Scale Testing (efficacy/feasibility)
  - System Manufacture/Assembly (currently)
  - Equipment Leasing & Reagent Supply
  - Technical Support/Training
  - Sublicensing to Geographic Zone Applicators
- Zone & Specialty Use Applicators
  - Project-Based Full-scale Use (Operations)
  - Personnel Training by trained and approved trainers



*All licensed technologies are patent(s)/patent(s) pending in the US and abroad*

# YB Technologies, LLC

Anacortes, WA

## Treatment Technologies:

- AMEOX® Technology - PFAS destruction for fluid concentrates, brines, and spent GAC
- MBT™ Treatment Technology: Leachable PFAS fixation/sequestration; heavy metals in soils/solid waste
- Advanced Neutralization™ (AN™): Acidity treatment & heavy metal/Rare Earth Element removal for mine water and landfill leachate pretreatment
- Tri-axial Mixing Reactor: Water treatment process component for water/media contact interactions
- In-Mine Closure™ (In-Mine Closure): Abandoned/legacy and active mine closure (AMD prevention)



*Technologies are patented/patent(s)-pending in the US and abroad*

# AMEOX<sup>®</sup> Destroys PFAS in

- Granular Activated Carbon for its Rejuvenation and Re-use
- Aqueous Fluid Concentrates
- Brines and Other Fluids

**...producing Fluoride and Carbon Dioxide**





# AMEOX® Full-Scale Dual Commercial Transportable Unit 2022 - current



# AMEOX™ Treatment Technology Defined

## Electrochemical Oxidation

- Boron-Doped Diamond electrode surfaces on Niobium substrate
- Minimizes over-potential for water cleavage and maximizes electron transfer
- Dimensionally stable/non-sacrificial

## Cavitation

- Cavitation is the formation and collapse of propagating bubbles in liquid
- Bubbles have internal temperature increases to thousands of deg-Kelvin and pressure to hundreds of bars
- Water vapor, oxygen and other constituents, are dissociated inside a bubble and yield oxidants such as ·OH, singlet O, ozone, peroxides, and others.

## Sonolysis

- Ultrasonic energy creates propagating cavitation bubbles in water and generates oxidants
- Recognized technology for the destruction of organic pollutants in water
- Treatment results in the destruction of the C – F PFAS bonds
- PFAS molecules are reduced to fluoride ions and CO<sub>2</sub>



# AMEOX® Treatment Technology Components

- Electrochemical Oxidation (Boron-Doped Diamond Electrodes)
- Cavitation (multiple forms)
- Sonolysis
- Intimate robust contact static mixing
- Oxidant Regeneration
- Closed-loop AMEOX Fluid Processing (No discharge/re-useable)
- In-media (carbon) & In-processor (desorbed) PFAS Destruction
- Oxidant Latency (**short-term half-life issues with hydroxyl radicals**)



# AMEOX® Technology Oxidants

## Oxidation Potential for Common Oxidants

<u>Oxidant</u>	<u>Oxidation Potential (V)</u>
Fluorine (F <sub>2</sub> )	3.0
Hydroxyl radical – acidic pH (•OH)	2.8
Sulfate radical (•SO <sub>4</sub> <sup>-</sup> )	2.6
Singlet (atomic) Oxygen (•O)	2.4
Ozone (O <sub>3</sub> )	2.1
Persulfate (S <sub>2</sub> O <sub>5</sub> <sup>-</sup> )	2.1
Hydroxyl radical – neutral pH (•OH)	1.8
Peroxymonosulfate (HSO <sub>5</sub> <sup>-</sup> )	1.8
Hydrogen Peroxide (H <sub>2</sub> O <sub>2</sub> )	1.8
Carbonate radical (•CO <sub>3</sub> <sup>-</sup> )	1.8
Perhydroxyl radical (HO <sub>2</sub> •)	1.7
Percarbonate (as Sodium percarbonate)	1.6
Chlorine dioxide (ClO <sub>2</sub> )	1.5
Chlorine (Cl <sub>2</sub> )	1.4
Oxygen (O <sub>2</sub> )	1.2
Hypochlorous Acid (HOCl)	0.95

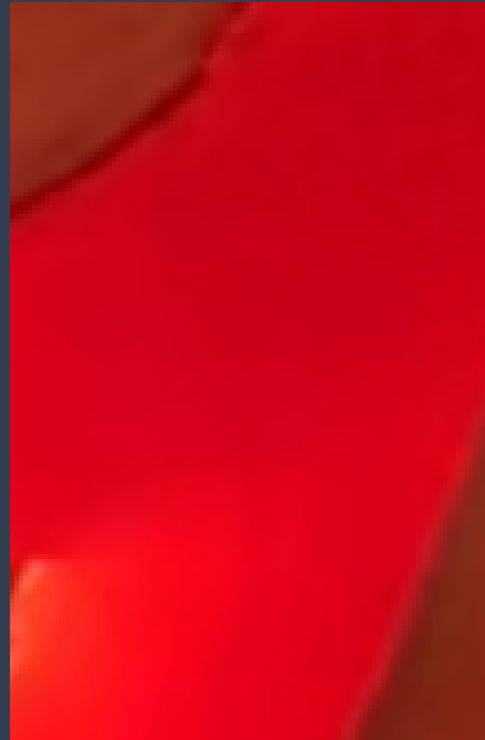




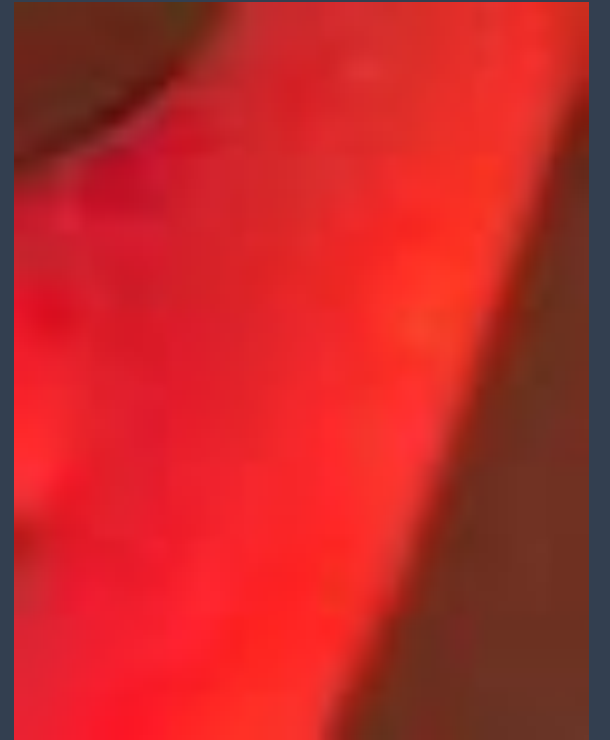
# AMEOX™ Contact Static Mixing Reactor Dye Test



Inadequate Mixing



Corrected Mixing



# AMEOX® Technology – How It Works:

For Granular Activated Carbon (GAC):

- AMEOX equipment attaches to carbon vessel containing PFAS-spent GAC
- AMEOX Fluid recirculates: Equipment through carbon bed in vessel back to equipment
- AMEOX Fluid delivers powerful oxidants to PFAS on GAC where PFAS is destroyed

For brines and fluid concentrates:

- AMEOX equipment attaches to a tank containing PFAS-laden brine/concentrate
- Fluid is circulated through AMEOX equipment back to tank to destroy PFAS
- AMEOX equipment generates powerful oxidants which destroy PFAS as fluid circulates



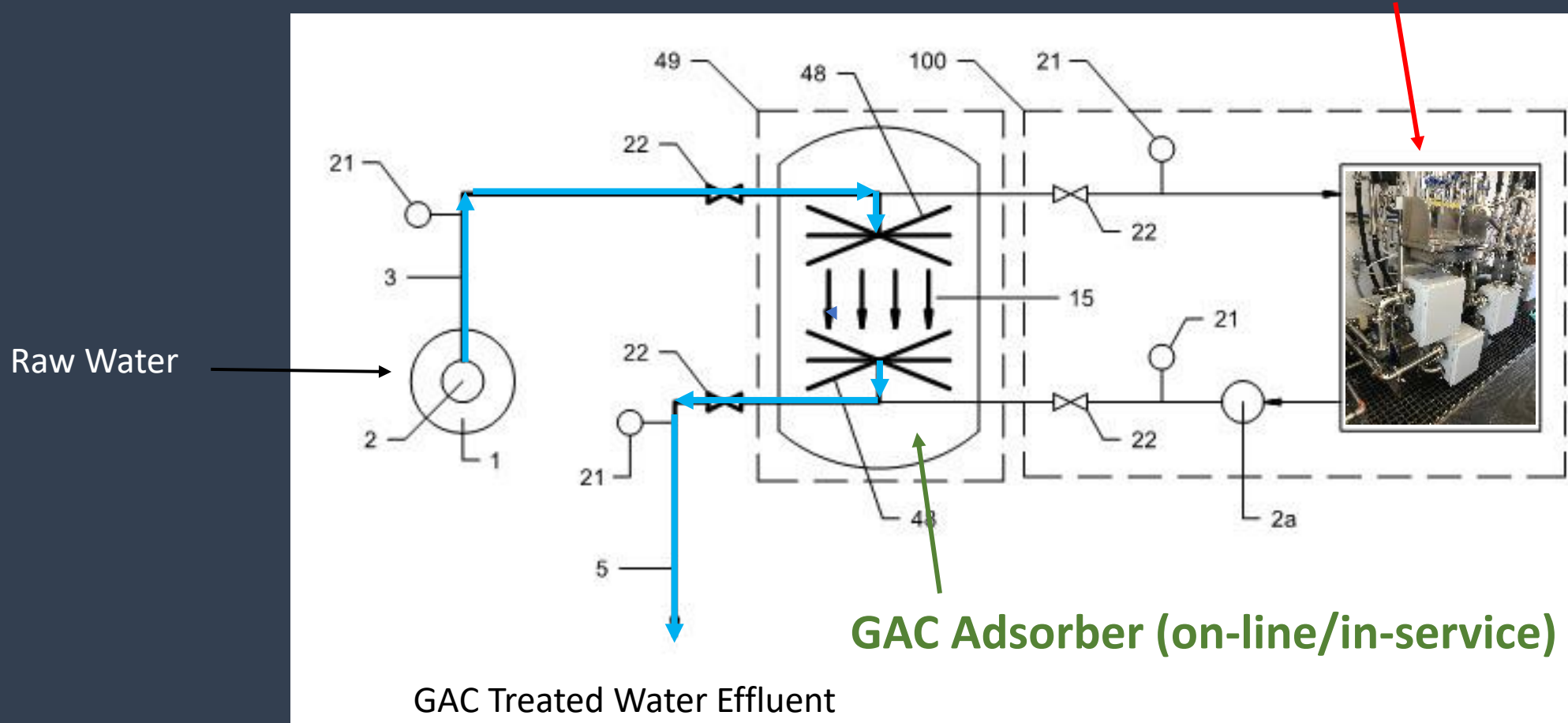
# 2022 AMEOX Commercial Scale Dual Unit



Treatment System within a Conex Box allows for mobility

# On-site AMEOX Spent GAC Processing

AMEOX™ System (off-line)



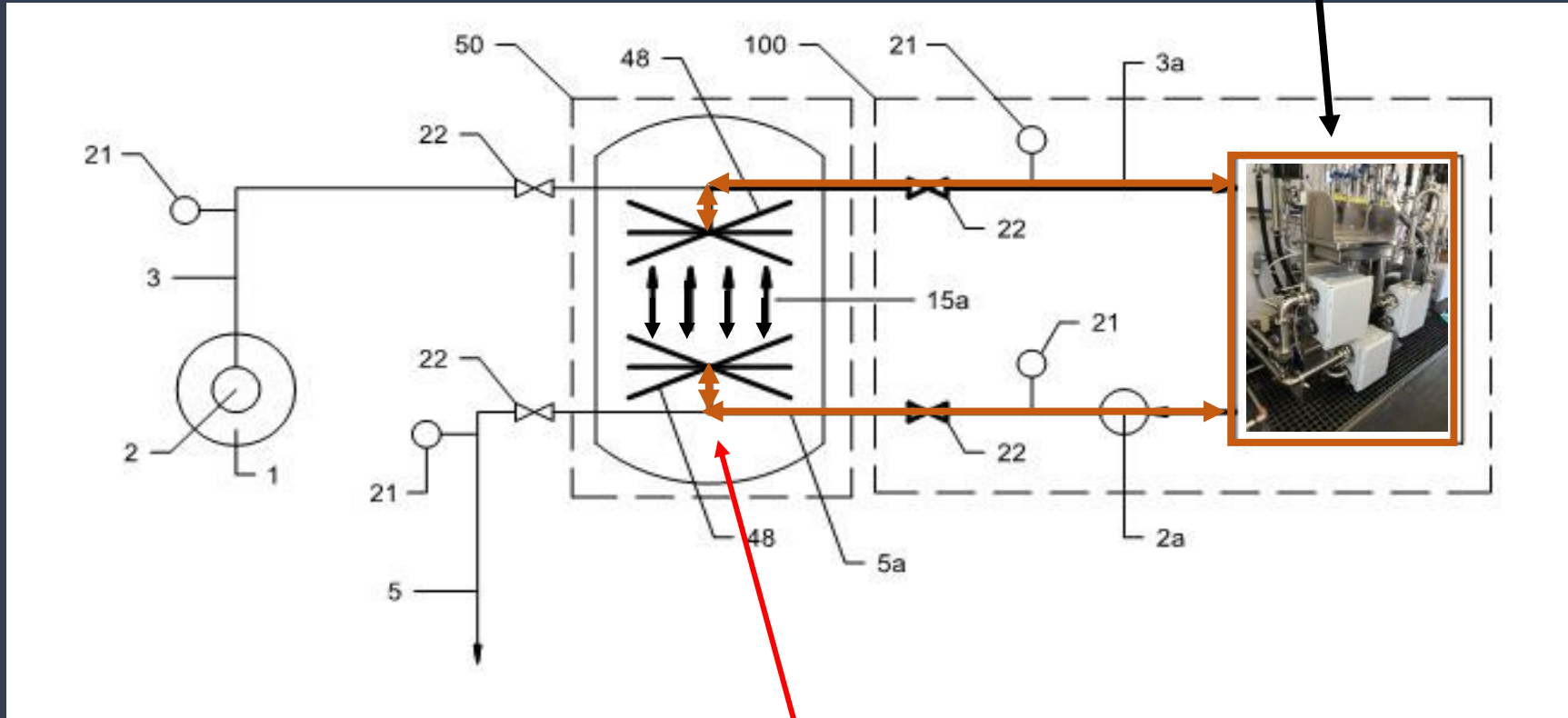
**SEE OUR PFAS DESTRUCTION VIDEO:** [https://www.youtube.com/watch?v=L1jl7L7\\_rxo](https://www.youtube.com/watch?v=L1jl7L7_rxo)





# On-site AMEOX Spent GAC Processing

**AMEOX™ System (on-line/in-service)**



**GAC Adsorber (off-line/out of service/in treatment)**

**SEE OUR PFAS DESTRUCTION VIDEO:** [https://www.youtube.com/watch?v=L1jl7L7\\_rxo](https://www.youtube.com/watch?v=L1jl7L7_rxo)

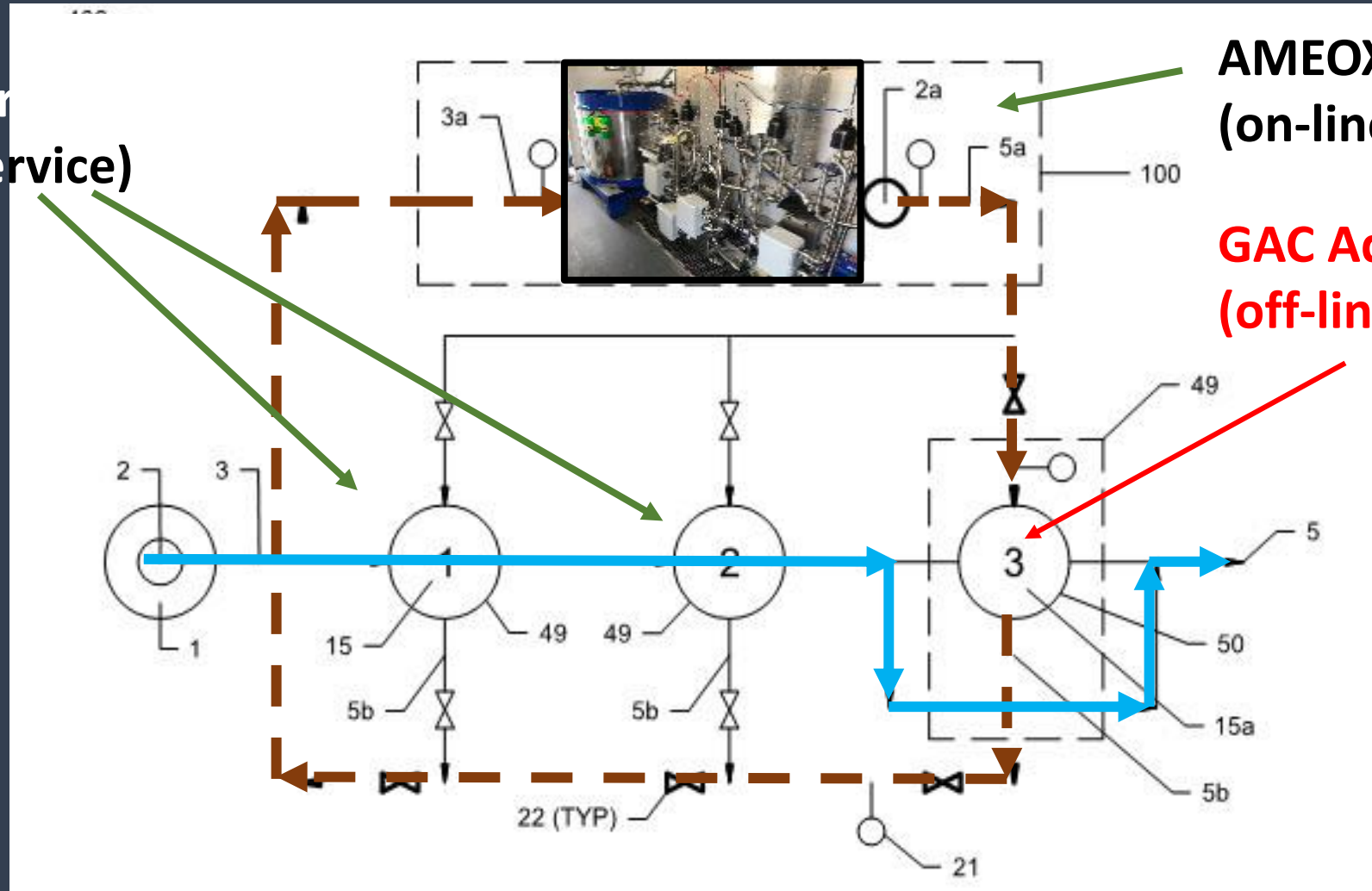




# On-site AMEOX Spent GAC Processing – Absorber Train

GAC Adsorber  
(on-line/in-service)

Raw Water Source



AMEOX™ System  
(on-line/in-service)

GAC Adsorber  
(off-line/in-treatment)

GAC Treat  
Water Effluent



# AMEOX® Dual Commercial Unit in Operation 2022- current



Dual Unit in 20' Shipping Container



1000# GAC  
Absorber Reactor



2000# GAC  
Absorber Reactor



# AMEOX GAC Processing Data

## AMEOX Treated GAC

### Returned to Service – Landfill Leachate Treatment

Wyoming, MI

June - October 2022

As of October 19, 2022:

- **Use #1:** Originally used on PFAS groundwater pump & treat system >>AMEOX treated (data on right >>>>>)
- **Use #2:** Landfill leachate decant >> AMEOX treated
- **Use #3:** Landfill leachate decant



			UNTREATED	Treated
			ALS	ALS
			06272022-1015	07252022-1020
Elapsed Treatment Time (Hrs.)			0	37.6
Matrix:			GAC	GAC
Supplemental Processing:				
			Total PFAS	Total PFAS
PFAS Telomers	Chemical Name	C Atoms	(ng/Kg - dry wt.)	(ng/Kg - dry wt.)
PFBA	Perfluorobutanoic Acid	C4	4,100	1,700
PFPeA	Perfluoropentanoic Acid	C5	4,600	1,700
PFHxA	Perfluorohexanoic Acid	C6	<29,000	4,000
PFHpA	Perfluoroheptanoic Acid	C7	<29,000	2,400
<b>PFOA</b>	<b>Perfluorooctanoic Acid</b>	<b>C8</b>	<b>47,000</b>	<b>8,400</b>
PFNA	Perfluorononanoic Acid	C9	560	200
PFDA	Perfluorodecanoic Acid	C10	610	<250
PFUnA	Perfluoroundecanoic Acid	C11	<290	<250
PFDoA	Perfluorododecanoic Acid	C12	<290	<250
PFTriA	Perfluorotridecanoic Acid	C13	<290	<250
PFTeA	Perfluortetradecanoic Acid	C14	<290	<250
PFBS	Perfluorobutanesulfonic Acid	C4	31,000	3,100
PFPeS	Perfluoropentanesulfonic Acid	C5	8,700	830
PFHxS	Perfluorohexanesulfonic Acid	C6	59,000	4,200
PFHpS	Perfluoroheptanesulfonic Acid	C7	62,000	3,900
<b>PFOS</b>	<b>Perfluorooctanesulfonic Acid</b>	<b>C8</b>	<b>7,500,000</b>	<b>780,000</b>
PFNS	Perfluorononanesulfonic Acid	C9	1,300	<250
PFDS	Perfluorodecanesulfonic Acid	C10	<59	<51
FtSA 4:2	Fluorotelomer Sulfonic Acid 4:2	C6	<290	<250
FtSA 6:2	Fluorotelomer Sulfonic Acid 6:2	C8	3,400	<250
FtSA 8:2	Fluorotelomer Sulfonic Acid 8:2	C10	<290	<250
PFOSA	Perfluorooctanesulfonamide	C8	67	<51
N-EtFOSSA	N-Ethylperfluorooctanesulfonic	C12	<290	<250
N-MeFOSAA	N-Methylperfluorooctanesulfonice	C11	<290	<250
F-53BMin	11Cl-Pf3OUds	C10	<59	<51
DONA	4,8-Doxa-3H-perfluorononanoic Acid	C8	<59	<51
F-53BMaj	9Cl-PF3ONS	C8	<59	<51
HFPO-DA	Hexafluoropropylene	C3	<290	<250
% Moisture	% Moisture		58	55
<b>Summation of PFOA and PFOS:</b>			<b>7,547,000</b>	<b>788,400</b>
<b>Summation of PFAS Telomers:</b>			<b>7,722,337</b>	<b>810,430</b>
Lab testing by ALS, Holland, MI				-89.6%
				-89.5%



# AMEOX<sup>®</sup> Dual Commercial Unit in Operation (January 2024)



10,000# Trailered GAC Absorbers (2 on left)  
AMEOX Dual Unit in 20' Shipping Container (rt.)



AMEOX GAC Processing (-8°F)



# EVENT #1: AMEOX GAC Processing Data – 10K# Lead Cell

Winter/Spring 2024

>80% PFAS  
Destruction

High Level  
Metals Removal

Rejuvenated  
GAC for Re-Use  
May 2024

GAC and Absorber				Source:	CONFIDENTIAL 10,000# GAC - Absorber Cell 1 (Lead)		
				Sample Type:	Untreated PFAS GAC	AMEOX Treated	
(2) 10k lbs. carbon cells on semi trailer Lead Cell 1 (9,000 lbs.) - Located at front of trailer Lag Cell 2 (9,000 lbs.) - Located at rear of trailer				Analytes:	PFAS Totals/ Metals	PFAS Totals/ Metals	
				Matrix:	GAC	GAC	
				Phase:	CHARACTERIZATION	Post Treatment	
				Sample ID:	11302023-1445	02082024W-0830	
				Method:	E-537 Mod	E-537 Mod	% Reduction
				Units (unless noted):	(ng/Kg-dry wt.)	(ng/Kg-dry wt.)	
Telomere	Chemical Name	CAS No.	C Atoms	Result	Result		
FtSA 6:2	Fluorotelomer Sulfonic Acid 6:2	27619-97-2	C8	<1800	3,900	ND	
PFBS	Perfluorobutanesulfonic Acid	375-73-5	C4	11,000	3,800	-65.5%	
PFBA	Perfluorobutanoic Acid	375-22-4	C4	40,000	26,000	-35.0%	
PFHpS	Perfluoroheptanesulfonic Acid	375-92-8	C7	15,000	4,000	-73.3%	
PFHpA	Perfluoroheptanoic Acid	375-85-9	C7	15,000	5,000	-66.7%	
PFHxS	Perfluorohexanesulfonic Acid	355-46-4	C6	37,000	11,000	-70.3%	
PFHxA	Perfluorohexanoic Acid	307-24-4	C6	14,000	7,100	-49.3%	
PFNA	Perfluorononanoic Acid	375-95-1	C9	6,300	2,400	-61.9%	
PFOS	Perfluorooctanesulfonic Acid	1763-23-1	C8	520,000	74,000	-85.8%	
PFOA	Perfluorooctanoic Acid	335-67-1	C8	29,000	12,000	-58.6%	
PFPeS	Perfluoropentanesulfonic Acid	2706-91-4	C5	8,800	3,000	-65.9%	
PFPeA	Perfluoropentanoic Acid	2706-90-3	C5	8,400	5,000	-40.5%	
HFPO-DA	Hexafluoropropylene oxide dimer acid	13252-13-6	C6	<1800	<2100		
PFecHS	Perfluoro-4-ethylcyclohexanesulfonic Acid	646-83-3	C8	880,000	140,000	-84.1%	
Al	Aluminum		mg/Kg	1,600	400	-75.0%	
Ca	Calcium		mg/Kg	3,800	98	-97.4%	
Fe	Iron		mg/Kg	6,800	3,700	-45.6%	
Mn	Manganese		mg/Kg	430	NA	-98.1%	
% Moisture					45	53	
Summation of PFOA and PFOS:					549,000	86,000	-84.3%
* Summation of PFOA, PFOS, PFNA, PFHxS, PFBS, HFPO-DA (GenX):					603,300	103,200	-82.9%
Summation of PFAS Telomere Totals:					1,573,500	289,500	-81.6%

Non-detected PFAS telomeres not shown





# EVENT #1: AMEOX GAC Processing Data – 10K# Lag Cell

Winter/Spring 2024

80% PFAS  
Destruction

High Level  
Metals Removal

Rejuvenated  
GAC for Re-Use  
May 2024

GAC and Absorber				CONFIDENTIAL 10,000# GAC - Absorber Cell 2 (Lag)																																																																																														
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430,000	79,000	-81.6%																																																																																																
473,800	93,200	-80.3%																																																																																																
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Non-detected PFAS telomeres not shown

# EVENT #2: AMEOX GAC Processing System Summer 2024

*“Data from the AMEOX-treated GAC indicated approximately 80% of the total PFAS (dry-wt.) was destroyed, and foulant removal including Al, Ca, Fe, and Mn ranged from 37% to over 98%. AMEOX processed GAC returned to PFAS water treatment service outperformed the original virgin GAC in adsorption capacity.”*

**Ramboll Americas Engineering Solutions, Inc.**

**RAMBOLL**



# EVENT #2: AMEOX GAC Processing Data – 10K# Lead Cell

Summer 2024

>80% PFAS Destruction

High Level  
Metals Removal

Rejuvenated GAC for Re-  
Use  
Fall 2024

Lead Cell - GAC and Absorber				Sample Type:	Untreated PFAS GAC	Treated R2	
2 x 10k lbs. carbon vessels on treatment semi trailer Lead Cell 1 (~<9,000 lbs.) - Located at front of trailer				Analytes:	PFAS Totals/ Metals	PFAS Totals/ Metals	
				Matrix:	GAC	GAC	
				Sample Source:	Event 2 - Lead Cell 1	Event 2 - Lead Cell 1	
				Method:	E-537 Mod	E-537 Mod	
				Units (unless noted):	(ng/Kg-dry wt.)	(ng/Kg-dry wt.)	
Telomere	Chemical Name	CAS No.	C Atoms	Result	Result	% Reduction	
FSA 4:2	Fluorotelomer Sulfonic Acid 4:2	757124-72-4	C6	<1500	<2100		
FSA 6:2	Fluorotelomer Sulfonic Acid 6:2	27619-97-2	C8	<1500	9,500	533.3%	
FSA 8:2	Fluorotelomer Sulfonic Acid 8:2	39108-34-4	C10	<1500	<2100		
FSA 10:2	Fluorotelomer Sulfonic Acid 10:2	120226-60-0	C12	<1500	<2100		
<b>PFBS</b>	<b>Perfluorobutanesulfonic Acid</b>	<b>375-73-5</b>	<b>C4</b>	<b>5,300</b>	<b>17,000</b>	<b>220.8%</b>	
<b>PFBA</b>	<b>Perfluorobutanoic Acid</b>	<b>375-22-4</b>	<b>C4</b>	<b>11,000</b>	<b>2,600</b>	<b>-76.4%</b>	
PFDS	Perfluorododecanesulfonic Acid	335-77-3	C10	<1500	<2100		
PFDA	Perfluorodecanoic Acid	335-76-2	C10	<1500	<2100		
PFDoS	Perfluorododecanesulfonic Acid	79780-39-5	C12	<1500	<2100		
PFDOA	Perfluorododecanoic Acid	307-55-1	C12	<1500	<2100		
PFHpS	Perfluoroheptanesulfonic Acid	375-92-8	C7	8,800	2,400	-72.7%	
PFHpA	Perfluoroheptanoic Acid	375-85-9	C7	7,800	<2100		
PFHxDA	Perfluorohexadecanoic Acid	67905-19-5	C16	<1500	<2100		
<b>PFHxS</b>	<b>Perfluorohexanesulfonic Acid</b>	<b>355-46-4</b>	<b>C6</b>	<b>17,000</b>	<b>19,000</b>	<b>11.8%</b>	
<b>PFHxA</b>	<b>Perfluorohexanoic Acid</b>	<b>307-24-4</b>	<b>C6</b>	<b>6,300</b>	<b>&lt;2100</b>		
PFNS	Perfluorononanesulfonic Acid	68259-12-1	C9	1,500	<2100		
<b>PFNA</b>	<b>Perfluorononanoic Acid</b>	<b>375-95-1</b>	<b>C9</b>	<b>4,700</b>	<b>&lt;2100</b>	<b>&gt; -55.3%</b>	
PFODA	Perfluorooctadecanoic Acid	16517-11-6	C18	<1500	<2100		
PFOSA	Perfluorooctanesulfonamide	754-91-6	C8	<1500	<2100		
<b>PFOS</b>	<b>Perfluorooctanesulfonic Acid</b>	<b>1763-23-1</b>	<b>C8</b>	<b>580,000</b>	<b>110,000</b>	<b>-81.0%</b>	
<b>PFOA</b>	<b>Perfluorooctanoic Acid</b>	<b>335-67-1</b>	<b>C8</b>	<b>15,000</b>	<b>&lt;2100</b>	<b>&gt; -86.0%</b>	
<b>PFPeS</b>	<b>Perfluoropentanesulfonic Acid</b>	<b>2706-91-4</b>	<b>C5</b>	<b>4,100</b>	<b>3,400</b>	<b>-17.1%</b>	
<b>PFPeA</b>	<b>Perfluoropentanoic Acid</b>	<b>2706-90-3</b>	<b>C5</b>	<b>4,200</b>	<b>&lt;2100</b>	<b>&gt; -50.0%</b>	
PFTeA	Perfluortetradecanoic Acid	376-06-7	C14	<1500	<2100		
PFTrDA	Perfluorotridecanoic Acid	72629-94-8	C13	<1500	<2100		
PFUnA	Perfluoroundecanoic Acid	2058-94-8	C11	<1500	<2100		
	N-ethylperfluoro-1-octanesulfonamide	936109-40-9	C10	<1500	<2100		
N-EtFOSSA	N-Ethylperfluorooctanesulfonamidoacetic Acid	2991-50-6	C12	<1500	<2100		
	N-Ethylperfluorooctanesulfonamidoethanol	N/A	C12	<1500	<2100		
N-MeFOSSA	N-methylperfluoro-1-octanesulfonamide	31506-32-8	C9	<1500	<2100		
N-MeFOSSAA	N-Methylperfluorooctanesulfonamidoacetic Acid	2355-31-9	C11	<1500	<2100		
N-MeFOSE	N-Methylperfluorooctanesulfonamidoethanol	24448-09-7	C11	<1500	<2100		
<b>HFPO-DA</b>	<b>Hexafluoropropylene oxide dimer acid</b>	<b>13252-13-6</b>	<b>C6</b>	<b>7,600</b>	<b>&lt;2100</b>	<b>&gt; - 72.4%</b>	
ADONA	4,8-Dioxa-3H-perfluorononanoic Acid	919005-14-4	C8	<1500	<2100		
F-53BMin	11Cl-PF3OUds	763051-92-9	C10	<1500	<2100		
F-53BMaj	9Cl-PF3ONS	756426-58-1	C8	<1500	<2100		
<b>PfEcHS</b>	<b>Perfluoro-4-ethylcyclohexanesulfonic Acid</b>	<b>646-83-3</b>	<b>C8</b>	<b>680,000</b>	<b>83,000</b>	<b>-87.8%</b>	
PFBSA	Perfluorobutylsulfonamide	30334-69-1	C4	<1500	<2100		
PFHxSA	Perfluorohexanesulfonamide	41997-13-1	C6	<1500	<2100		
<b>Parameter</b>							
<b>Al</b>	<b>Aluminum</b>	SW3050B	mg/Kg	770	10,000	1199%	
<b>Ca</b>	<b>Calcium</b>	SW3050B	mg/Kg	2,300	1,500	-35%	
<b>Fe</b>	<b>Iron</b>	SW3050B	mg/Kg	4,300	3,600	-16%	
<b>Mn</b>	<b>Manganese</b>	SW3050B	mg/Kg	150	33.0	-78%	
<b>% Moisture</b>		SW3550C	%	34	52		
				Summation of PFOA and PFOS:	595,000	110,000	-81.5%
				* Summation of PFOA, PFOS, PFNA, PFHxS, PFBS, HFPO-DA):	622,000	<150,200	> - 75.9%
				Summation of PFAS Telomere Totals:	1,348,000	220,400	-83.6%



# EVENT #2: AMEOX GAC Processing Data – 10K# Lag Cell

Summer 2024

>80% PFAS Destruction

High Level Metals Removal

Rejuvenated GAC for Re-Use  
Fall 2024

Lag Cell - GAC and Absorber				Sample Type:	Untreated	Treated	
				Analytes:	PFAS Totals / Metals		PFAS Totals / Metals
				Matrix:	GAC		GAC
				Sample Source:	Event 2 - Lag Cell 2		Event 2 - Lag Cell 2
				Method:	E-537 Mod	E-537 Mod	% Reduction
				Units (unless noted):	(ng/Kg-dry wt.)	(ng/Kg-dry wt.)	
Telomere	Chemical Name	CAS No.	CAtoms	Result	Result		
FtSA 4:2	Fluorotelomer Sulfonic Acid 4:2	757124-72-4	C6	<2000	<21800		
FtSA 6:2	Fluorotelomer Sulfonic Acid 6:2	27619-97-2	C8	<2000	<2270		
FtSA 8:2	Fluorotelomer Sulfonic Acid 8:2	99108-34-4	C10	<2000	<22300		
FtSA 10:2	Fluorotelomer Sulfonic Acid 10:2	120226-60-0	C12	<2000	<2310		
PFBS	Perfluorobutanesulfonic Acid	375-73-5	C4	6,100	<2110		> -65.4%
PFBA	Perfluorobutanoic Acid	375-22-4	C4	13,000	3,160		-75.7%
PFDS	Perfluorodecanesulfonic Acid	335-77-3	C10	<2000	<2310		
PFDA	Perfluorodecanoic Acid	335-76-2	C10	<2000	<2390		
PFDoS	Perfluorododecanesulfonic Acid	79780-39-5	C12	<2000	<22,600		
PFDoA	Perfluorododecanoic Acid	307-55-1	C12	<2000	<2390		
PFHpS	Perfluoroheptanesulfonic Acid	375-92-8	C7	14,000	<2280		> -83.7%
PFHpA	Perfluoroheptanoic Acid	375-85-9	C7	9,300	<2390		> -74.3%
PFHxDA	Perfluorohexadecanoic Acid	67905-19-5	C16	<2000	<2390		
PFHxS	Perfluorohexanesulfonic Acid	355-46-4	C6	27,000	2,930		-89.1%
PFHxA	Perfluorohexanoic Acid	307-24-4	C6	9,300	<2390		> -74.3%
PFNS	Perfluorononanesulfonic Acid	68259-12-1	C9	<2000	<2390		
PFNA	Perfluorononanoic Acid	375-95-1	C9	7,400	<2390		> -67.7%
PFODA	Perfluorooctadecanoic Acid	16517-11-6	C18	<2000	<2390		
PFOSAm	Perfluorooctane sulfonamide	754-91-6	C8	<2000	<2390		
PFOS	Perfluorooctanesulfonic Acid	1763-23-1	C8	710,000	104,000		-85.4%
PFOA	Perfluorooctanoic Acid	335-67-1	C8	23,000	3,450		-85.0%
PFPeS	Perfluoropentanesulfonic Acid	2706-91-4	C5	6,200	<2390		> -61.5%
PFPeA	Perfluoropentanoic Acid	2706-90-3	C5	5,400	<2390		> -55.7%
PFTDA	Perfluorotetradecanoic Acid	376-06-7	C14	<2000	<23,300		
PFTDA	Perfluorotridecanoic Acid	72629-94-8	C13	<2000	<2390		
PFUnDA	Perfluoroundecanoic Acid	2058-94-8	C11	<2000	<2390		
EtFOSAm	N-ethylperfluoro-1-octane sulfonamide	936109-40-9	C10	<2000	<2390		
NETFOSSA	N-Ethylperfluorooctanesulfonamidoacetic Acid	2991-50-6	C12	<2000	<2390		
EtFOSE	N-Ethylperfluorooctane sulfonamidoethanol	N/A	C12	<2000	<23,300		
N-MeFOSA	N-methylperfluorooctane sulfonamide	31506-32-8	C9	<2000	<23,300		
N-MeFOSAA	N-Methylperfluorooctane sulfonamido acetic Acid	2355-31-9	C11	<2000	<23,300		
N-MeFOSE	N-Methylperfluorooctane sulfonamidoethanol	24448-09-7	C11	<2000	2,390		
HFPO-DA	Hexafluoropropylene oxide dimer acid	13252-13-6	C6	15,000	2,830		-81.1%
DONA	4,8-Dioxa-3H-perfluorononanoic Acid	919005-14-4	C8	<2000	<2250		
F-53BMIn	11Cl-PF3OUds	763051-92-9	C10	<2000	<2250		
F-53BMaj	9-Cl-PF3ONS	756426-58-1	C8	<2000	<2230		
PFecHS	Perfluoro-4-ethylcyclohexanesulfonic Acid	646-83-3	C8	780,000	178,000		-77.2%
PFBSA	Perfluorobutylsulfonamide	30334-69-1	C4	<2000	<2390		
PFHxSA	Perfluorohexanesulfonamide	41997-13-1	C6	<2000	<2390		
<b>Parameter</b>							
Al	Aluminum	SW3050B	mg/Kg	770	741		-3.8%
Ca	Calcium	SW3050B	mg/Kg	840	82		-90.2%
Fe	Iron	SW3050B	mg/Kg	4,500	894		-80.1%
Mn	Manganese	SW3050B	mg/Kg	26	5.95		-77.1%
% Moisture		SW3550C	%	50	57.1		
				Summation of PFOA and PFOS:	733,000	107,450	-85.3%
				* Summation of PFOA, PFOS, PFNA, PFHxS, PFBS, HFPO-DA:	773,500	<114,300	> -85.2%
				Summation of PFAS Telomere Totals:	1,625,700	296,760	-81.7%



## AMEOX Performance Indicators:

1. Commercial GAC re-use record with performance enhanced over new GAC
2. Decrease in PFAS telomere concentrations
3. Carbon dioxide – low level (ppt) real-time monitoring not viable

### 4. *Fluoride* increase in AMEOX Re-circulation Process Fluid:

<u>Program</u>	<u>Parameter</u>	<u>Pre-AMEOX</u>	<u>Post-AMEOX</u>
2020 - Prineville, OR	F <sup>-</sup> (mg/L)	<0.1	<b>0.21</b>
2022 - Wyoming, MI	F <sup>-</sup> (mg/L)	0.8	<b>7.48</b>
2023-24 Wyoming, MI	F <sup>-</sup> (mg/L)	0.8	<b>41.6</b>

### 5. *Iron/Calcium* in GAC

2022 – Wyoming, MI	Iron (mg/Kg)	18,000	<b>2,500</b>
2024 – Wyoming, MI	Calcium (mg/Kg)	9,300	<b>2,200</b>





# AMEOX® Fluid

## Residual Solids (2024)

- Post-AMEOX Processing of 2 x 10K lbs. GAC Absorbers
- PFAS Telemers all N.D. in recovered solids
- Metals removed from exhausted GAC during AMEOX processing
- Consolidated solids suitable for dewatering and subsequent disposal in a RCRA Subtitle D Non-Hazardous waste landfill

AMEOX Fluids: Solids 2 x 10,000 lbs. GAC Absorber Processing Consolidated Solids after precipitation, settling, and decant from AMEOX Fluid			Source: Sample Type:	AMEOX Fluid: Post-GAC Processing Settled Solids
Telomere	Chemical Name	CAS No.	Sample ID: Method: C Atoms	Post GAC AMEOX Treatment CT Solids - (9/2024) E-537 Mod Result (ug/Kg-dry wt.)
F-53BMin	11Cl-PF3OUds	763051-92-9	C10	<33.0
FtSA 8:2	Fluorotelomer Sulfonic Acid 8:2	39108-34-4	C10	<33.5
FtSA 10:2	Fluorotelomer Sulfonic Acid 10:2	120226-60-0	C12	<33.8
FtSA 4:2	Fluorotelomer Sulfonic Acid 4:2	757124-72-4	C6	<32.7
FtSA 6:2	Fluorotelomer Sulfonic Acid 6:2	27619-97-2	C8	<33.2
FTCA 7:3	2H,2H,3H,3H-Perfluorodecanoic acid 7:3	812-70-4	C10	<35.0
FTCA 3:3	2H,2H,3H,3H-Perfluorohexanoic acid 3:3	356-02-5	C6	<35.0
FTCA 5:3	2H,2H,3H,3H-Perfluorooctanoic acid 5:3	914637-49-3	C8	<35.0
DONA	4,8-Dioxa-3H-perfluorononanoic Acid	919005-14-4	C8	<33.0
F-53BMaj	9-Cl-PF3ONS	756426-58-1	C8	<32.6
HFPO-DA	Hexafluoropropylene oxide dimer acid	13252-13-6	C6	<35.0
EtFOSAm	N-Ethylperfluorooctane sulfonamide	936109-40-9	C10	<35.0
NEtFOSSA	N-Ethylperfluorooctane sulfonamido acetic Acid	2991-50-6	C12	<35.0
EtFOSE	N-Ethylperfluorooctane sulfonamido ethanol	1691-99-2	C12	<35.0
N-MeFOSA	N-methylperfluorooctane sulfonamide	31506-32-8	C9	<35.0
N-MeFOSAA	N-Methylperfluorooctane sulfonamido acetic Acid	2355-31-9	C11	<35.0
N-MeFOSE	N-Methylperfluorooctane sulfonamido ethanol	24448-09-7	C11	<35.0
PFecHS	Perfluoro-4-ethylcyclohexanesulfonic Acid	646-83-3	C8	<35.0
PFBS	Perfluorobutanesulfonic Acid	375-73-5	C4	<31.0
PFBA	Perfluorobutanoic Acid	375-22-4	C4	<35.0
PFBSA	Perfluorobutylsulfonamide	30334-69-1	C4	<35.0
PFDS	Perfluorodecane sulfonic Acid	335-77-3	C10	<33.8
PFDA	Perfluorodecanoic Acid	335-76-2	C10	<35.0
PFDoS	Perfluorododecane sulfonic Acid	79780-39-5	C12	>33.9
PFDoA	Perfluorododecanoic Acid	307-55-1	C12	<35.0
PFHpS	Perfluoroheptane sulfonic Acid	375-92-8	C7	<33.3
PFHpA	Perfluoroheptanoic Acid	375-85-9	C7	<35.0
PFHxDA	Perfluorohexadecanoic Acid	67905-19-5	C16	<35.0
PFHxS	Perfluorohexane sulfonic Acid	355-46-4	C6	<35.0
PFHxSA	Perfluorohexanesulfonamide	41997-13-1	C6	<35.0
PFHxA	Perfluorohexanoic Acid	307-24-4	C6	<35.0
PFNS	Perfluorononane sulfonic Acid	68259-12-1	C9	<35.0
PFNA	Perfluorononanoic Acid	375-95-1	C9	<35.0
PFODA	Perfluorooctadecanoic Acid	16517-11-6	C18	<35.0
PFOSAm	Perfluorooctane sulfonamide	754-91-6	C8	<35.0
PFOS	Perfluorooctane sulfonic Acid	1763-23-1	C8	<35.0
PFOA	Perfluorooctanoic Acid	335-67-1	C8	<35.0
PFPeS	Perfluoropentane sulfonic Acid	2706-91-4	C5	<35.0
PFPeA	Perfluoropentanoic Acid	2706-90-3	C5	<35.0
PFTDA	Perfluortetradecanoic Acid	376-06-7	C14	<35.0
PFTrDA	Perfluorortridecanoic Acid	72629-94-8	C13	<35.0
PFUnDA	Perfluoroundecanoic Acid	2058-94-8	C11	<35.0
<b>Parameter</b>				
Al	Aluminum	SW3050B	mg/Kg-dry	15,300
Ca	Calcium	SW3050B	mg/Kg-dry	126,000
Fe	Iron	SW3050B	mg/Kg-dry	8,130
Mn	Manganese	SW3050B	mg/Kg-dry	16,900
% Moisture		SW3550C	%	97.1
Summation of PFOA and PFOS:				<70 (ND)
* Summation of PFOA, PFOS, PFNA, PFHxS, PFBS, HFPO-DA:				<206 (ND)
Summation of PFAS Telomere Totals:				<1448.8 (ND)



# AMEOX<sup>®</sup> Technology for PFAS Destruction (August 2022)

## Beneficial Iron Removal from Spent GAC

Untreated GAC:

18,000 mg/Kg

Treated GAC:

2,500 mg/Kg



# AMEOX<sup>®</sup> Fluid

## Re-Use Data (2023-2024)

- PFAS “ND” in AMEOX fluid & settled solids.
- Fluoride generated during PFAS destruction and removed from fluid with Al, Ca, Fe, and Mn foulants.
- F<sup>-</sup> background in city makeup water for AMEOX fluid was 0.8 mg/L. Foulant metals sourced from carbon.
- AMEOX fluid returned for re-use to destroy PFAS in GAC.
- Fluid processed for metals and fluoride multiple times since spring 2023.
- Solids accumulated since start of processing.



AMEOX Fluid and Solids				Source:	Cone Tank	
AMEOX fluid = Supernate after solids settling and decant Solids = Solids after settling and decant of AMEOX fluid Solids accumulated for nearly 1 year				Sample Type:	Supernate Decant	Settled Solids
				Analytes:	PFAS Totals/ Metals	PFAS Totals/ Metals
				Matrix:	AMEOX Fluid	Solids
				Lab:	ALS	ALS
				PFAS Method:	E-537 Mod (ng/L)	E-537 Mod (ug/Kg-dry wt.)
				Units (unless noted):		
Telomere	Chemical Name	CAS No.	C Atoms	Result	Result	
FtSA 4:2	Fluorotelomer Sulfonic Acid 4:2	757124-72-4	C6	<5.0	<35	
FtSA 6:2	Fluorotelomer Sulfonic Acid 6:2	27619-97-2	C8	<5.0	<35	
FtSA 8:2	Fluorotelomer Sulfonic Acid 8:2	39108-34-4	C10	<5.0	<35	
FtSA 10:2	Fluorotelomer Sulfonic Acid 10:2	120226-60-0	C12	<5.0	<35	
<b>PFBS</b>	<b>Perfluorobutanesulfonic Acid</b>	<b>375-73-5</b>	<b>C4</b>	<b>&lt;5.0</b>	<b>&lt;35</b>	
<b>PFBA</b>	<b>Perfluorobutanoic Acid</b>	<b>375-22-4</b>	<b>C4</b>	<b>&lt;5.0</b>	<b>&lt;35</b>	
PFDS	Perfluorodecanesulfonic Acid	335-77-3	C10	<5.0	<35	
PFDA	Perfluorodecanoic Acid	335-76-2	C10	<5.0	<35	
PFDoS	Perfluorododecanesulfonic Acid	79780-39-5	C12	<5.0	<35	
PFDoA	Perfluorododecanoic Acid	307-55-1	C12	<5.0	<35	
PFHpS	Perfluoroheptanesulfonic Acid	375-92-8	C7	<5.0	<35	
PFHpA	Perfluoroheptanoic Acid	375-85-9	C7	<5.0	<35	
PFHxDA	Perfluorohexadecanoic Acid	67905-19-5	C16	<5.0	<35	
<b>PFHxS</b>	<b>Perfluorohexanesulfonic Acid</b>	<b>355-46-4</b>	<b>C6</b>	<b>&lt;5.0</b>	<b>&lt;35</b>	
<b>PFHxA</b>	<b>Perfluorohexanoic Acid</b>	<b>307-24-4</b>	<b>C6</b>	<b>&lt;5.0</b>	<b>&lt;35</b>	
PFNS	Perfluorononanesulfonic Acid	68259-12-1	C9	<5.0	<35	
<b>PFNA</b>	<b>Perfluorononanoic Acid</b>	<b>375-95-1</b>	<b>C9</b>	<b>&lt;5.0</b>	<b>&lt;35</b>	
PFODA	Perfluorooctadecanoic Acid	16517-11-6	C18	<5.0	<35	
PFOSA	Perfluorooctanesulfonamide	754-91-6	C8	<5.0	<35	
<b>PFOS</b>	<b>Perfluorooctanesulfonic Acid</b>	<b>1763-23-1</b>	<b>C8</b>	<b>&lt;5.0</b>	<b>&lt;35</b>	
PFOA	Perfluorooctanoic Acid	335-67-1	C8	<5.0	<35	
<b>PFPeS</b>	<b>Perfluoropentanesulfonic Acid</b>	<b>2706-91-4</b>	<b>C5</b>	<b>&lt;5.0</b>	<b>&lt;35</b>	
<b>PFPeA</b>	<b>Perfluoropentanoic Acid</b>	<b>2706-90-3</b>	<b>C5</b>	<b>&lt;5.0</b>	<b>&lt;35</b>	
PFTeA	Perfluortetradecanoic Acid	376-06-7	C14	<5.0	<35	
PFTrDA	Perfluorortridecanoic Acid	72629-94-8	C13	<5.0	<35	
PFUnA	Perfluoroundecanoic Acid	2058-94-8	C11	<5.0	<35	
	N-ethylperfluoro-1-octanesulfonamide	936109-40-9	C10	<5.0	<35	
N-EtFOSSA	N-Ethylperfluorooctanesulfonamidoacetic Acid	2991-50-6	C12	<5.0	<35	
	N-Ethylperfluorooctanesulfonamidoethanol	N/A	C12	<5.0	<35	
N-MeFOSA	N-methylperfluoro-1-octanesulfonamide	31506-32-8	C9	<5.0	<35	
N-MeFOSAA	N-Methylperfluorooctanesulfonamidoacetic Acid	2355-31-9	C11	<5.0	<35	
N-MeFOSE	N-Methylperfluorooctanesulfonamidoethanol	24448-09-7	C11	<5.0	<35	
<b>HFPO-DA</b>	<b>Hexafluoropropylene oxide dimer acid</b>	<b>13252-13-6</b>	<b>C6</b>	<b>&lt;5.0</b>	<b>&lt;35</b>	
ADONA	4,8-Dioxa-3H-perfluorononanoic Acid	919005-14-4	C8	<5.0	<35	
F-53BMin	11CI-PF3OUs	763051-92-9	C10	<5.0	<35	
F-53BMaj	9CI-PF3ONS	756426-58-1	C8	<5.0	<35	
<b>PFecHS</b>	<b>Perfluoro-4-ethylcyclohexanesulfonic Acid</b>	<b>646-83-3</b>	<b>C8</b>	<b>&lt;5.0</b>	<b>&lt;35</b>	
PFBSA	Perfluorobutylsulfonamide	30334-69-1	C4	<5.0	<35	
PFHxSA	Perfluorohexanesulfonamide	41997-13-1	C6	<5.0	<35	
<b>Parameter</b>						
F	Fluoride (SW9056A)			<0.40 mg/L	41.6	
Al	Aluminum (SW3015A / 3050B)			0.11 mg/L	15,000	
Ca	Calcium (SW3015A / 3050B)			330 mg/L	100,000	
Fe	Iron (SW3015A / 3050B)			0.26 mg/L	42,000	
Mn	Manganese (SW3050B)			N/A	20,000	
% Moisture	Moisture (SW3550C)		% of sample:	N/A	97	
				Summation of PFOA and PFOS:	Non-Detect	Non-Detect
				* Summation of PFOA, PFOS, PFNA, PFHxS, PFBS, HFPO-DA (GenX):	Non-Detect	Non-Detect
				Summation of PFAS Telomere Totals:	Non-Detect	Non-Detect

# AMEOX<sup>®</sup> Treated Carbon

## Inorganic Parameters (2023-24)

<u>Parameter</u>	<u>Untreated GAC</u>	<u>AMEOX Treated GAC</u>
Iron	18,000	2500
Aluminum	5100	2600
Calcium	9300	2200
Manganese	2800	400
Alkalinity (total as CaCO <sub>3</sub> )	250	ND (<170)

\* All units as mg/Kg (dry wt.)





# AMEOX GAC Processing Data

PFOA & PFOS to <DL's

Bed-Depth Treatment Data

Industrial Waste Recovery

Wyoming, MI

July 2022



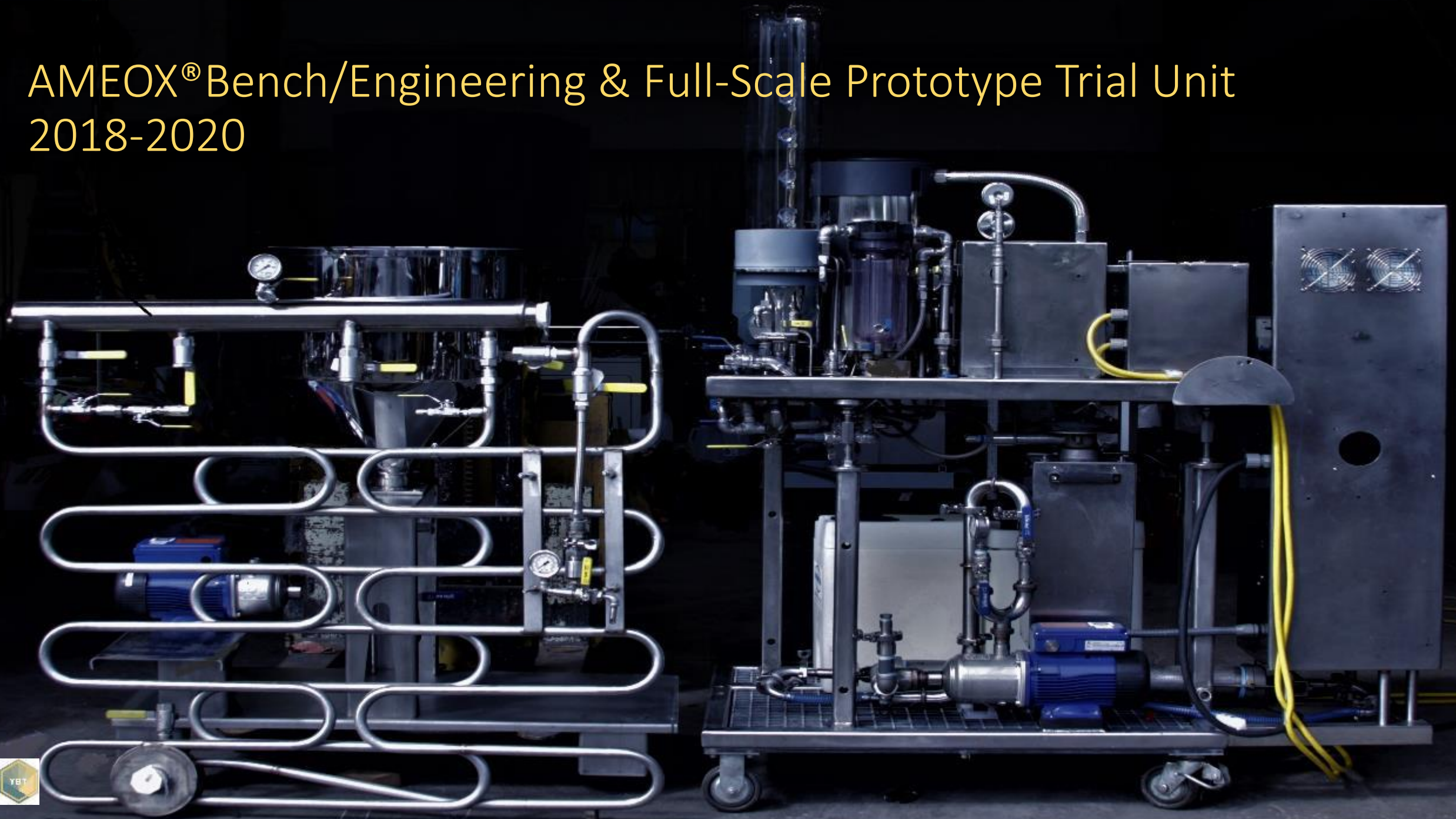
			Treatment Status:	UNTREATED	TREATED	TREATED
			Elapsed Treatment Time (Hrs.):	T = 0	17.3	17.3
			Matrix:	GAC	GAC	GAC
			1000# GAC Bed Depth Sample Location:	Vertical Composite	Top 1/2	Bottom 1/2
PFAS Telomers	Chemical Name	C Atoms		TOTAL PFAS (ng/Kg - dry wt.)	TOTAL PFAS (ng/Kg - dry wt.)	TOTAL PFAS (ng/Kg - dry wt.)
PFBA	Perfluorobutanoic Acid	C4		8,000	240	230
PFPeA	Perfluoropentanoic Acid	C5		7,300	<120	<210
PFHxA	Perfluorohexanoic Acid	C6		<22,000	<120	<210
PFHpA	Perfluoroheptanoic Acid	C7		<22,000	<120	<210
<b>PFOA</b>	<b>Perfluorooctanoic Acid</b>	<b>C8</b>		<b>24,000</b>	<b>&lt;25</b>	<b>&lt;43</b>
PFNA	Perfluorononanoic Acid	C9		870	<25	<43
PFDA	Perfluorodecanoic Acid	C10		220	<120	<210
PFUnA	Perfluoroundecanoic Acid	C11		220	<120	<210
PFDoA	Perfluorododenoic Acid	C12		220	<120	<210
PFTriA	Perfluorotridecanoic Acid	C13		220	<120	<210
PFTeA	Perfluortetradecanoic Acid	C14		220	<120	<210
PFBS	Perfluorobutanesulfonic Acid	C4		7,300	30	<43
PFPeS	Perfluoropentanesulfonic Acid	C5		2,900	<25	<43
PFHxS	Perfluorohexanesulfonic Acid	C6		<22,000	<120	<210
PFHpS	Perfluoroheptanesulfonic Acid	C7		5,400	<120	<210
<b>PFOS</b>	<b>Perfluorooctanesulfonic Acid</b>	<b>C8</b>		<b>1,600,000</b>	<b>&lt;25</b>	<b>&lt;43</b>
PFNS	Perfluorononanesulfonic Acid	C9		1,000	<120	<210
PFDS	Perfluorodecanesulfonic Acid	C10		<44	<25	<43
FtSA 4:2	Fluorotelomer Sulfonic Acid 4:2	C6		<220	<120	<210
FtSA 6:2	Fluorotelomer Sulfonic Acid 6:2	C8		8,400	<120	<210
FtSA 8:2	Fluorotelomer Sulfonic Acid 8:2	C10		<220	<120	<210
PFOSA	Perfluorooctanesulfonamide	C8		<44	<25	<43
N-EtFOSSA	N-Ethylperfluorooctanesulfonic	C12		<220	<120	<210
N-MeFOSAA	N-Methylperfluorooctanesulfonic	C11		<220	<120	<210
F-53BMin	11Cl-Pf3OUds	C10		<44	<25	<43
DONA	4,8-Doxa-3H-perfluorononanoic Acid	C8		<44	<25	<43
F-53BMaj	9Cl-PF3ONS	C8		<44	<25	<210
HFPO-DA	Hexafluoropropylene	C3		<220	<120	<210
% Moisture	% Moisture			45	43	40
<b>Summation of PFOA and PFOS:</b>				<b>1,624,000</b>	<b>ND</b>	<b>ND</b>
<b>Summation of PFAS Telomers:</b>				<b>1,666,270</b>	<b>270</b>	<b>230</b>
Lab testing by ALS, Holland, MI					>99%	>99%



AMEOX® Dual Commercial Unit in Operation  
Waste Recovery Systems, Inc.  
Wyoming, MI (since May 2022)



# AMEOX<sup>®</sup> Bench/Engineering & Full-Scale Prototype Trial Unit 2018-2020





# AMEOX<sup>®</sup> Bench-Scale Treatability Unit (2020)



# On-site AMEOX<sup>®</sup> Processing (February 2024): Reverse Osmosis Reject Concentrate



RO Concentrate Treatment  
AMEOX System Setup



Treatment at Start



Treatment at 30 minutes





# AMEOX Processing Data – Reverse Osmosis Reject Concentrate

## February 2024

Viability Study data shows RO-90% fluid concentrate is suitable for full-scale optimized processing via the AMEOX technology.

				AMEOX® Technology Viability Study for PFAS Destruction: Minnesota Landfill Leachate VSEP RO 90% Recovery Reject Concentrate		
<b>Processed Fluid</b>				<b>Untreated RO Reject (90%)</b>		<b>AMEOX Treated RO Reject (90%)</b>
VSEP Concentrate				<b>PFAS Totals/ Metals</b>		<b>PFAS Totals/ Metals</b>
7/5/23				<b>RO Concentrate Liquid</b>		<b>RO Concentrate Liquid</b>
90% Recovery				<b>CHARACTERIZATION</b>		
~20 gallons				<b>T = 0</b>		<b>T = 16</b>
				<b>AMEOX Reactor</b>		<b>AMEOX Reactor</b>
<b>Source:</b>				<b>02092024-RO-90%</b>		<b>02102024-RO-90%</b>
<b>Sample Type:</b>				<b>ALS</b>		<b>ALS</b>
<b>Analytes:</b>				<b>E-537 Mod / SW3015A</b>		<b>E-537 Mod / SW3015A</b>
<b>Matrix:</b>				<b>(ng/L)</b>		<b>(ng/L)</b>
<b>Treatability Phase:</b>						<b>% Reduction</b>
<b>Elapsed Hours Treated:</b>						
<b>Sample Source:</b>						
<b>Sample ID:</b>						
<b>Lab:</b>						
<b>Method(s):</b>						
<b>Units (unless noted):</b>						
<b>Telomere</b>	<b>Chemical Name</b>	<b>CAS No.</b>	<b>C Atoms</b>	<b>Result</b>	<b>Result</b>	
FtSA 6:2	Fluorotelomer Sulfonic Acid 6:2	27619-97-2	C8	910	35	-96.2%
PFBS	Perfluorobutanesulfonic Acid	375-73-5	C4	6,700	3,900	-41.8%
PFBA	Perfluorobutanoic Acid	375-22-4	C4	11,000	9,000	-18.2%
PFHpA	Perfluoroheptanoic Acid	375-85-9	C7	5,900	500	-91.5%
PFHxS	Perfluorohexanesulfonic Acid	355-46-4	C6	4,200	240	-94.3%
PFHxA	Perfluorohexanoic Acid	307-24-4	C6	22,000	17,000	-22.7%
PFNA	Perfluorononanoic Acid	375-95-1	C9	270	<25	-90.7%
PFOS	Perfluorooctanesulfonic Acid	1763-23-1	C8	580	<25	-95.7%
PFOA	Perfluorooctanoic Acid	335-67-1	C8	7,500	210	-97.2%
PFPeS	Perfluoropentanesulfonic Acid	2706-91-4	C5	250	120	-52.0%
PFPeA	Perfluoropentanoic Acid	2706-90-3	C5	11,000	9,200	-16.4%
HFPO-DA	Hexafluoropropylene oxide dimer acid	13252-13-6	C6	<99	<25	
PFecHS	Perfluoro-4-ethylcyclohexanesulfonic Acid	646-83-3	C8	140	<25	-82.1%
PFBSA	Perfluorobutylsulfonamide	30334-69-1	C4	300	190	-36.7%
<b>Summation of PFOA and PFOS:</b>				<b>8,080</b>	<b>210</b>	<b>97.4%</b>
<b>* Summation of PFOA, PFOS, PFNA, PFHxS, PFBS, HFPO-DA (GenX):</b>				<b>19,250</b>	<b>4,350</b>	<b>77.4%</b>
<b>Summation of PFAS Telomere Totals:</b>				<b>70,750</b>	<b>40,437</b>	<b>42.8%</b>

### NOTES:

- \* March 14, 2023 - Proposed PFAS National Primary Drinking Water Regulation - (6) PFAS: <https://www.epa.gov/sdwa/and-polyfluoroalkyl-substances-pfas>
- VSEP Concentrate - 90% Recovery Landfill Leachate RO concentrate was processed February 9-10, 2024 using an AMEOX Single Unit
- AMEOX processing followed a generic treatment regime design for PFOA + PFOS and using data provided by customer for Total PFAS of up to 10ug/L
- Data indicate AMEOX processing conditions and RO fluid characteristics caused dissolution of metal oxide deposits accumulated on internal surfaces of AMEOX equipment from previous GAC processing.
- PFAS cross-over contamination not evaluated. No equipment blank was tested as fluid run through system was RO concentrate (vs. clean water) and any carryover from piping surfaces would be insignificant for this viability study.
- Viability data confirms the AMEOX technology is able to reduce the level of PFAS in 90% RO concentrate and indicates process optimization is feasible.



# AMEOX<sup>®</sup> Technology Fact Sheet

- **>60-65% PFAS destruction in GAC creates effective capacity in GAC for its reuse**
- >99% destruction increases operating costs if desired.
- Electric Power Platform (240V/1-ph/50A service with propane heat)
- Robust Vortex Mixing (no motors/moving parts)
- Stainless Steel Construction
- Skid-mounted (or trailer/shipping container housing)
- <1000 to 5000-10,000+# lb. media absorber reactor processing
- Re-usable AMEOX Fluid (oxidant carrier) with no discharge
- 2 pumps w/VFD per AMEOX Single Unit system (1 hp ea.)



# AMEOX<sup>®</sup> Technology Key Benefits

- 1) Destroys PFAS in GAC ending the PFAS cycle
- 2) No discharge of AMEOX fluid. Sustainably recycled
- 3) Low power (240V/1-ph/<50A)
- 4) Ambient temperature (100oF), low pressure (<80psi)
- 5) Carbon re-use estimate: ~7X. 4X to date for landfill leachate
- 6) Small 4' x 12' footprint. Portable or fixed site application
- 7) Flexibly accommodates range of GAC adsorber/reactor sizes



# AMEOX® Full-scale GAC Processing Examples – Summary Table

<u>Spent Carbon Source</u>	<u>Parameters</u>	<u>Untreated GAC (ng/Kg-dry)</u>	<u>AMEOX Treated GAC (ng/Kg-dry)</u>	<u>Destruction (%)</u>	<u>Comment</u>
Groundwater Remediation	PFOA + PFOS	3,313,000	485,500	-85.3%	Full-scale commercial. Re-used for Landfill Leachate Treatment
	Total PFAS	3,343,780	513,250	-84.7%	
Landfill Leachate Treatment	PFOA + PFOS	7,547,000	788,400	-89.6%	Full-scale commercial. Treated 2nd time after 2nd reuse. Original from groundwater remediation. Use #3 for Landfill Leachate
	Total PFAS	7,722,337	810,430	-89.5%	
Mfg Plant Basement AFFF Fire-Fight Water, Detroit, MI	PFOA + PFOS	254,200	86,500	-66.0%	Full-scale commercial. Re-used for Landfill Leachate Treatment
	Total PFAS	271,513	93,716	-65.5%	
Landfill Leachate Treatment (Top 1/2 GAC)	PFOA + PFOS	1,624,000	ND	>99%	Full-scale Commercial. Re-used for Landfill Leachate Treatment.
	Total PFAS	1,666,270	270	>99%	
Landfill Leachate Treatment (Bottom 1/2 GAC) * Top+Bottom composited pre-treat	PFOA + PFOS	1,624,000*	ND	>99%	
	Total PFAS	1,666,270*	230	>99%	
AFFF Pit Water GAC	PFOA + PFOS	1,900,000	ND	97%	Full-scale Commercial
	Total PFAS	2,408,550	ND	97%	
State of MI Superfund GW Remediation Site. Production Plated Plastics, Richland, MI	PFOA + PFOS	22,937,500	6,841,000	-70.2%	Full-scale Pilot (2020)
	Total PFAS	23,249,398	6,995,500	-69.9%	
City of Ann Arbor, MI DW Treatment Plant GAC, Ann Arbor, MI	PFOA + PFOS	104,448	6,800	-93.5%	Full-scale Pilot (2020)
	Total PFAS	379,446	31,400	-91.7%	

**NOTES:**

- 1) GAC with PFAS destruction at >65% was suitable for re-use to treat landfill leachate for sanitary discharge. Destruction to DL not necessary for re-use.
- 2) Iron, calcium, aluminum, manganese and alkalinity removal data not shown
- 3) No physical degradation of GAC due to AMEOX processing (EPS Analytical)



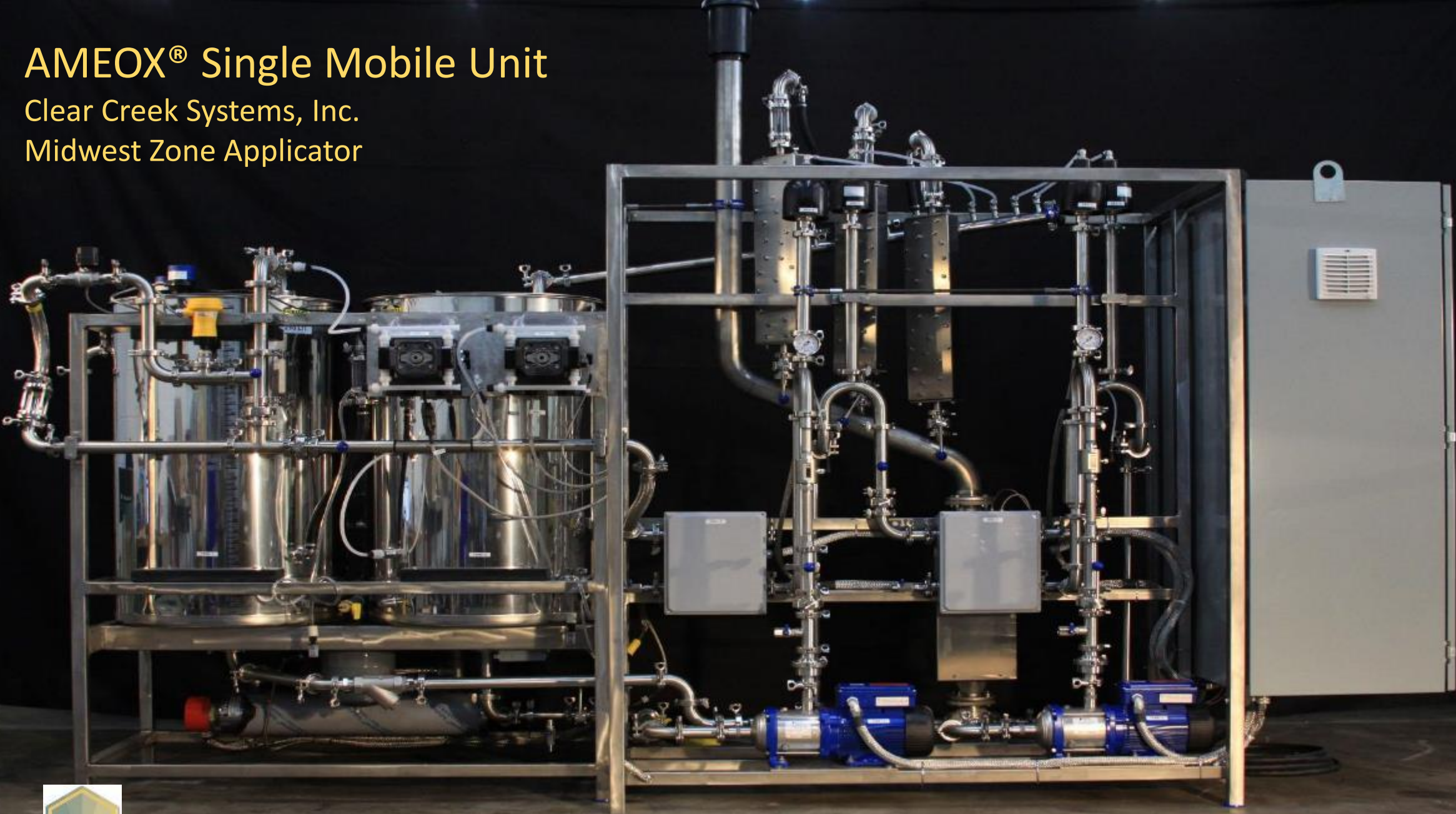
Analytical data by ALS Global, Holland, MI for State of MI (EGLE/MPART) PFAS Telomere List (see individual data slides – attached)



# AMEOX® Single Mobile Unit

Clear Creek Systems, Inc.

Midwest Zone Applicator



# AMEOX® Single Unit Mobile Systems (Fabrication to Readiness Testing)

YB Technologies, LLC

Anacortes, WA

August 2023





# AMEOX – Process Control & Cost Variables:

1. GAC Quality (Coal, Coconut/Other, Re-agglomerated, Virgin, Service, etc.)
2. Water Chemistry (treated by GAC and used for AMEOX Fluid)
3. Operating Temperature (35°F – 125°F) and Pressure Range (atmospheric to 80-85 psi)
4. Local power rates, power intensity, frequency, and modulation
5. PFAS Concentrations/Mass of GAC per Reactor Batch
6. GAC end-use/performance objectives (Reuse Reactivation, Internment, Subsequent Processing)
7. 3-5<sup>+</sup>-yr AMEOX Equipment Life Expectancy (w/pumps @ 2-3 years with servicing)

**GAC characterization data and process objectives required for cost estimate**



# AMEOX® Destruction of PFAS

## Alternative Application Potentials





# AMEOX Technology Application Potential

## Alternate Adsorption/Absorption Media

- Ion Exchange Resin
- Organoclay
- Other

### Issues/Concerns:

- Media structural stability?
- Base material type?
- PFAS destruction?
- Disposal or Re-use?



# AMEOX Technology Application Potential

AFFF Fire Suppression System Retrofits

On-site/In-Line PFAS Destruction









# QUESTIONS?

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